

L 55890-65 EWT(1)/EWT(m)/FCC/ENG(v)/EEC-4/EWP(j)/EEC(t)/EWA(h) Pe-4/Po-4/
Fe-5/Pq-4/Pac-2/Peb/Pt-4 RM/GW

ACCESSION NR: AR5014437

UR/0169/65/000/005/B010/B010
651.508

49
46
8

SOURCE: Ref. zh. Geofizika, Abs. 5B82

AUTHOR: Avdeyev, A.I.; Fridzon, M.B.; Kalinin, V.N.

TITLE: The protection of temperature sensors against radiation

CITED SOURCE: Sb. 150 let Meteorol. observ. Kazansk. un-ta. Kazan', Kazansk. un-t,
1963, 200-212

TOPIC TAGS: meteorological instrument, temperature sensor, stratosphere, anti-radiation coating, radiation error, silver passivation, silver reflectivity, lacquer coating, aluminum reflectivity

TRANSLATION: Silver applied to a polished base has the best reflecting properties of all the coverings used for the protection of stratospheric temperature sensors against radiation. However, silver is unstable under the influence of atmospheric factors. The authors have investigated several methods for maintaining a high reflectivity of silver by its passivation by chromic anhydride and by application of a protective coating. The reflectivity of samples stored for 15 to 20 days in the open air and under room conditions were determined in the wave-length interval 0.2-10 microns. The results of the

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measurements show that lacquers B-1 and AV-4, as well as silicon dioxide, decrease the reflectivity of silver and cannot be recommended for anti-radiation coatings. Passivation of the silver layer by chromic anhydride and the application of a methacrylic lacquer does not result in a decrease of the reflection coefficient and such processing provides long-term protection of the silver against atmospheric factors. The technical process which has been developed for the processing of the metal parts of the sensors makes possible a considerable reduction of the radiation error in measurements. In addition to the study of silver coatings, an investigation was made of the radiation characteristics of an aluminum layer. Aluminum has an absorption band in the region 0.6-0.9 microns in which about 30% of the solar energy falls. Although aluminum coatings have poorer reflective properties than silver coatings, the cheapness and stability of aluminum make its use highly promising for a number of meteorological instruments used on a large scale. M. Kaganov.

SUB CODE: ES ENCL: 00

COC
Card 2/2

L 55891-65 EWT(1)/FCC GW

ACCESSION NR: AR5014436

UR/0169/65/000/005/B010/B010
551.508.2

SOURCE: Ref. zh. Geofizika, Abs. 5B81

AUTHOR: Avdeyev, A.I.; Fridzon, M.B.; Kalinin, A.I.

TITLE: Some methods and results of experimental investigations of sensors of meteorological elements

CITED SOURCE: Ref. zh. Geofizika, Abs. 5B81

TOPIC TAGS: meteorological instrument, thermometer design, resistance thermometer,
radiation error, atmospheric temperature

TRANSLATION: Platinum resistance thermometers of both framework and openwork types have been developed. The thermometer is designed for balloon sounding of the atmosphere. It is manufactured from platinum wire 0.03-0.05 mm in diameter. The sensing element is attached by a capron filament to a silvered steel frame constructed of wire 2 mm in diameter. Tests have shown that at a height of about 3,000 m the radiation error has a value of about 0.3-0.4C and the thermal inertia is 0.2 sec. The thermometer readings are recorded by a measuring instrument, the basis of which is a bridge circuit

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ACCESSION NR: AR5014436

and a highly sensitive oscillograph with photorecording. A step-by-step switch, through a system of relays, ensures the alternate switching of the sensors and the control resistances. The sensitivity of the circuit is 3 μ amps per 1C. The interrogation rate is two sensors per second. The total weight of the measuring apparatus is 3.5 kg.

M. Kaganov

SUB CODE: ES ENCL: 00

coc
Card 2/2

BREDIKIS, Yu.I., kand. med. nauk; FRIDZON, M.G. (Moskva)

Use of electrical stimulation of the heart in Morgagni-Adams-Stokes syndrome developing against a background of severe diabetes mellitus. Probl. endok. i germ. 9 no.5:85-88 S-0'63
(MIRA 16:12)

1. Iz kliniki fakul'tetskoy khirurgii imeni S.I.Spasokukotskogo (dir. - akademik A.N. Bakulev) II Moskovskogo meditsinskogo instituta imeni N.I.Pirogova.

L 34828-65 EMT(1)/EPF(c)/EPF(n)-2/ Pr-4/Pu-4
ACCESSION NR: AP5007457

LJP(c) MM
S/0286/65/300/004/0077/0077

30
B

AUTHORS: Pashkovskiy, B. A.; Fridzon, M. G.

TITLE: Device for temperature measurement. Class 42, No. 168499

SOURCE: Byulleten' izobreteniy i tovarnykh znakov, no. 4, 1965, '77

TOPIC TAGS: temperature measurement, temperature gage, temperature sensitive element

ABSTRACT: This Author Certificate presents a device for temperature measurement containing a high-frequency oscillation generator and a tank circuit consisting of an inductance coil and capacitor, used as the temperature sensing element. For automation and continuity of measurements, the tank circuit is in the feedback circuit of the generator (see Fig. 1 on the Enclosure). Orig. art. has: 1 diagram.

ASSOCIATION: none

SUBMITTED: 01oct62

ENCL: 01

SUB CODE: TD, EC

NO REF SOV: 000

OTHER: 000

Card 1/2

"APPROVED FOR RELEASE: 06/13/2000

CIA-RDP86-00513R000513720017-5

AVDEYEV, A.I.; KALININ, V.N.; FRIDZON, M.V.

Protection of thermoreceivers from the thermal action of solar
radiation when measuring temperature at great heights. Trudy
TSAO no.41:86-90 '62.
(MIRA 16:10)

APPROVED FOR RELEASE: 06/13/2000

CIA-RDP86-00513R000513720017-5"

ZEFIROVA, G.S.; FRIDZON, R.G.

Pregnancy and labor in Addison's disease. Akush. i gir. no.1;
L45-146 '65.
(MIRA 18;10)

1. Kafedra endokrinologii (zav.- prof. Ye.A. Vasykova) Tsentral'no-
go instituta usovershenstvovaniya vrachey (dir.- M.D. Kovrigina)
i rodil'nyy dom No.25 (glavnyy vrach Ye.A. Sitnikova), Moskva.

FRIE, Fridrich

Nitrogen determination by the Kjeldahl method without distillation
(A contribution to nitrogen determination in plant and biological
material). Biologia 16 no.12:918-920 '61.

1. Biologicky ustav Slovenskej akademie vied, Oddelenie fyziologie
rastin v Bratislave.
(NITROGEN chemistry)

FRIEBOLE, L.

POLAND / Chemical Technology. Chemical Products and
Their Application. Instruments for Control
and Measurement and Automation.

H

Abs Jour: Ref Zhur-Khimiya, No 9, 1959, 31649.

Author : Friebol, E.

Inst : Not given.

Title : Drum Measuring Devices for Volumes of Liquids.

Orig Pub: Pomiary, automat., kontrola, 1957, 3, No 11,
429-431.

Abstract: The principle and structure of drum gauges (DG) for measuring volumes of liquids is described. General problems of the accuracy of measuring with the aid of DG, and the source and extent of errors, are examined. A number of graphs are introduced, portraying the influence, on the measuring accuracy, of the configuration

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REGULATORY

Country : Poland
Category :

H-3

Ass. Jour. :

46104

Aut. or : Friebel, E.
Institut. :
Title : Liquid Volume Drum-Meters. Part II.

Orig. Pub. : Pomiary, automat. kontrola, 1957, 3,
No 12, 470-473

Abstract : Extensive use of drum-meters (DM) was furthered by the basic characteristics of this apparatus: a) wide measurement-range of liquid flow rate, 50-12000 liter/hour; b) high accuracy ($\pm 0.5\%$) and sensitivity, which are retained over a great length of time for measurements ranging up to 1:100; c) dependable operation and the possibility of manufacturing the parts of DM which are in contact with the liquid, from corrosion resistant materials (ceramics). Hence DM are used to measure acids and alkalies, alcohols, benzene, gasoline, acetone, carbon disulfide, molasses, fruit juices, and also contaminated and viscous liquids. A description is

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Damping arrangement for a regulator.. S/196/62/000/012/015/016
E194/E155

stage of the cascade. Because of its voltage displacement the choke achieves the necessary phasing of rectification. The circuit can be used to operate with running times of about 15 seconds whilst turning a control handle through 90°.

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B

[Abstractor's note: Complete translation.]

Card 2/2

RYCHETSKY, Ladislav, inz.; FRIEBEL, Vilem

Securing the development of complex automation. Automatizace 7 no. 3:57-59 Mr '64.

1. Zavody prumyslove automatizace, Praha.

"APPROVED FOR RELEASE: 06/13/2000

CIA-RDP86-00513R000513720017-5

RYCHTER, Ladislav, inz.; FRIMBEN, Vojtěch

Use 17 automatic switches in me...ent circuits. Manual page 7
no.10; Supplement: 161-168 p. 164.

APPROVED FOR RELEASE: 06/13/2000

CIA-RDP86-00513R000513720017-5"

"APPROVED FOR RELEASE: 06/13/2000

CIA-RDP86-00513R000513720017-5

RYCHETSKY, L., inaz. FRIENDS, Vilem

First National Conference on Liquid Flow Control Devices.
Automatizace 8 no.1:6 Ja '65.

APPROVED FOR RELEASE: 06/13/2000

CIA-RDP86-00513R000513720017-5"

"APPROVED FOR RELEASE: 06/13/2000

CIA-RDP86-00513R000513720017-5

FRIEBEL, Vilem

New standardized series of orifice fittings. Automatizace 8 no.1:
11-14 Ja '65.

1. Zavody prumyslove automatizace National Enterprise, Prague.

APPROVED FOR RELEASE: 06/13/2000

CIA-RDP86-00513R000513720017-5"

FRIEBOVA, Z.; JELINEK, R.

Proliferation into the central nervous system of chick embryos.
III. Mitotic activity in the encephalic pouches between the 2d
and 6th day of incubation. Cesk. morf. 11 no.3:229-236 '63.

1. Zubni oddeleni polikliniky v Hlinsku, anatomicky ustav
Karlovych universit v Praze.
(CENTRAL NERVOUS SYSTEM) (CELL DIVISION)

KRYSL, J.; technicka spoluprace FRIEBOVA, Zd.

Filing roentgenological findings with the aid of edge punched cards. Cesk. rentgen 17 no.2;108-115 Mr '63.

1. Klinicka zakladna rtg-katedry UDL nemocnici v Praze na Bulovce,
prednosti MUDr. J. Slanina.
(PUNCHED CARD SYSTEMS) (RADIOGRAPHY)

JELINEK, R.; FRIEBOVA, Z.

Central nervous system proliferation in the chick embryo. IV.
Sborn. lek. 67 no.12:359-367 D ' 65.

1. Anatomicky ustanov fakulty všeobecného lékařství University
Karlových v Praze (prednosta - prof. MUDr. et RNDr. L. Borovanský,
DrSc.)

FRIED, A.; SRDINKA, V.; TAUBER, M.

Cholelithiasis and gastric chemistry. Gastroenterologia bohema 4
no.2-4:216-220 Oct 50. (CMLL 20:5)

1. J.Jessenius and Marie Curie Sanatorium (Head--Andrej Fried,
M.D.) of the Czechoslovak State Spa (General Director--R.Bures,
M.D.) in Karlove Vary.

FRIED, A.

KOVAROVIC, J.; FRIED, A.

Blood modifications following infectious hepatitis. Voj.zdrav. listy
19 no.11-12:266-267 Nov-Dec 50. (CIML 20:5)

FRIED, A.

Carbohydrates metabolism following gastric resection for carcinoma.
Sborn. patofysiolog. trav. vyz. 6 no. 4-6:298-299 Dec 1952. (CLML 24:1)

1. Karlovy Vary.

PRIED, Andrej

Viewpoint of the spa physician in regard to indications & contraindications of spa therapy at Karlovy Vary. Cas. lek. česk. 97 no.18:568-570
2 May 58.

1. Karlovy Vary, Lazenska lecebna Jesenius.
(MINERAL WATERS, ther. use
sulfur waters at Karlovy Vary, Czech., indic. & contraindic.
(Cz))

BANKI, Dezso; BARTHA, Jozsef; HEGEDUS, Jozsef, okleveles villamosmernok; TOTH, Otto; FRIED, Arnold; UNK, Janosne; FOLDEAK, Gabor; NIEWELT, Ferenc; KUCZOGI, Endre

Remarks about Aurel Felkai's entitled "Experiences with the operation of the Hungarian-manufactured heavy-current cables and lines." Villamossag 8 no.2-3:60-62 F-Mr '60.

1. Budapest Fovaros Elektromos Muvei vezeto mernoke (for Banki).
2. Lenin Kohaszati Muvek energia gyarreszlege fomernoke (for Bartha).
3. Orszagos Banyamuszaki Felugyeleseg (for Hegedus).
4. Borsodi Vegyi Kombinat foenergetikusa; Nehezipari Miniszterium
5. EM Nevezegyipari Foosztalya kepviselетеiben (for Toth).
6. EM Szereloiipari Tervezo Vallalat, Sztalinvaros (for Fried).
7. Magyar Asvanyolaj es Foldgazkiserleti Intezet (for Foldeak).
8. Villamosgep es Kabelgyar (for Niewelt).
9. Orszagos Villamosenergia Felugyelet (for Kuczogi).

Fried, Ervin.

~~Fried, Ervin.~~ Fields which can be represented as a quotient field of an integral domain properly contained in them. Eötvös. L. Tud.-Egy. Kiadv. Term.-Tud. Kar. Évk. 1952-53, 27-29 (1954). (Hungarian)

Turán asked the question: characterize the commutative fields which can be represented as the quotient field of an integral domain properly contained in the field. The author proves that the necessary and sufficient condition for a field to have this property is that it should not be an algebraic extension of a finite field. Several applications are discussed. [This paper appears also in a German version, Acta Sci. Math. Szeged 15 (1954), 143-144; MR 16, 992.]

P. Erdős (Haifa).

1 - F/W

200
1

P. Erdős

FRIEDE

Fried, E. Über als echte Quotientenkörper darstellbare
Körper. Acta Sci. Math. Szeged 15, 143-144 (1954).

A field K is representable as a true quotient field if K
contains a domain of integrity $I \neq K$ and K is the quotient
field of I . It is proved, without the use of the evaluation
theory, that K is representable as a true quotient field if
and only if K is not an algebraic extension field (finite or
infinite) of a finite field.

F. Kückemeister.

$I = F/W$

"APPROVED FOR RELEASE: 06/13/2000

CIA-RDP86-00513R000513720017-5

FRIED, E.

"Linear combinations of roots."
Kozlemenyei, Budapest, Vol 4, No 1, 1954, p. 155

SO: Eastern European Accessions List, Vol 3, No 10, Oct 1954, Lib. of Congress

APPROVED FOR RELEASE: 06/13/2000

CIA-RDP86-00513R000513720017-5"

6413:

1-FW
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2

Fried, Ervin, Algebraically closed fields as finite extensions. Mat. Lapok 7 (1956), 47-60. (Hungarian, Russian and English summaries)

Let A be an algebraically closed field and $V \subset A$ a proper subfield of A . Suppose that A is obtained from V by adjunction of a finite number of (a priori not necessarily algebraic) elements. Then V is a real closed field and $A = V(i)$. The proofs and formulations are modifications and slight improvements of known results. [See N. Bourbaki, Algèbre, chap. VI, Actualités Sci. Ind. no. 1179, Hermann, Paris, 1952; MR 14, 237; problems on pp. 47-48.]

Further it is shown: If F is a field and all irreducible polynomials over F have a degree less than a fixed integer, then either F is algebraically closed (and all irreducible polynomials are of degree ≤ 1) or F is real closed (and all irreducible polynomials are of degree ≤ 2).
St. Schwarz (Bratislava)

"APPROVED FOR RELEASE: 06/13/2000

CIA-RDP86-00513R000513720017-5

FRIED, Ervin, a matematikai tudomanyok kandidatusa

Open debate about Gyorgy Gratzer's dissertation for candidacy.
Mat kozl MTA 11 no.4:457-458 '61.

APPROVED FOR RELEASE: 06/13/2000

CIA-RDP86-00513R000513720017-5"

"APPROVED FOR RELEASE: 06/13/2000

CIA-RDP86-00513R000513720017-5

FRIED, Ervin, a matematikai tudomanyok kandidatusa

Open debate about Tamas E. Schmidt's dissertation for candidacy.
Mat kozl MTA 11 no.4:458-459 '61.

APPROVED FOR RELEASE: 06/13/2000

CIA-RDP86-00513R000513720017-5"

HAJOS, Gyorgy; KALMAR, Laszlo; SURANYI, Janos; TURAN, Pal; POSA, Lajos;
DE BRUJIN, N.G. (Amsterdam, Holland); SARKADI, Karoly; FRIED,
Ervin; WIEGANDT, Richard; ERDOS, Fal

Mathematical problems. Mat Lapok 12 no.3/4:253-258 '61.

- 1."Matematikai Lapok" szerkesztoje (for Hajos and Kalmar).
- 2."Matematikai Lapok" felelos szerkesztoje (for Turan).

"APPROVED FOR RELEASE: 06/13/2000

CIA-RDP86-00513R000513720017-5

FRIED, Ervin

Remarks on the introduction of complex numbers. Mat lapok 14 no.
1/2:103-106 '63.

APPROVED FOR RELEASE: 06/13/2000

CIA-RDP86-00513R000513720017-5"

CSASZAR, Akos; ERDOS, Pal; TURAN, Pal; KARTESZI, Ferenc; FRIED, Ervin;
WIEGANDT, Richard; CSIPSZER, Janos; KALMAR, Laszlo; KONCZ, Karoly;
MAJTHAY, Antal ; BOGDAN, Zoltan; HAJNA, Janos; HETYEI, Gabor;
SURANYI, Janos

Mathematical problems. Mat lapok 14 no.1/2:163-169 '63.

1. "Matematikai Lapok" felelos szerkesztoje (for Turan). 2. "Matematikai Lapok" szerkeszto bizottsagi tagja (for Kalmar),

"APPROVED FOR RELEASE: 06/13/2000

CIA-RDP86-00513R000513720017-5

CSASZAR, Akos; FRIED, Ervin; FUCHS, Laszlo; HAJOS, Gyorgy; RENYI, Alfred;
TURAN, Pal

Report on the 1962 Miklos Schweitzer Memorial Contest on
Mathematics. Mat lapok 14 no. 3/4:346-371 '63.

1. Editorial board member, "Matematikai Lapok" (for Hajos and
Renyi). 2. Managing editor, "Matematikai Lapok" (for Turan).

APPROVED FOR RELEASE: 06/13/2000

CIA-RDP86-00513R000513720017-5"

"APPROVED FOR RELEASE: 06/13/2000

CIA-RDP86-00513R000513720017-5

FRIED, Ervin

Short proof of the basic theorem of finite Abelian groups.
Mat lapok 15 no.1/3:225-227 '64

APPROVED FOR RELEASE: 06/13/2000

CIA-RDP86-00513R000513720017-5"

"APPROVED FOR RELEASE: 06/13/2000

CIA-RDP86-00513R000513720017-5

DEREVICI, V., ing.; FRIED, G.

Fastening of footwear soles by gluing. (Conclusion). Industria
usoara 3 no.12:491-495 D '56.

APPROVED FOR RELEASE: 06/13/2000

CIA-RDP86-00513R000513720017-5"

"APPROVED FOR RELEASE: 06/13/2000

CIA-RDP86-00513R000513720017-5

DEREVICI, V., ing.; FRIED, Gg.

Waste vegetable skins as substituting material for the manufacture of footwear heels. Industria usoara 3 no.2:58-59 F '56.

APPROVED FOR RELEASE: 06/13/2000

CIA-RDP86-00513R000513720017-5"

"APPROVED FOR RELEASE: 06/13/2000

CIA-RDP86-00513R000513720017-5

DEREVICI, V., ing.; FRIED, Gh.

Fastening of footwear soles by gluing. I. (To be contd.).
Industria usoara 3 no.11:447-452 N '56.

APPROVED FOR RELEASE: 06/13/2000

CIA-RDP86-00513R000513720017-5"

SZEP, Ivan; FRIED, Henrik; PINTER, Janos

Some manufacturing problems of germanium junction transistors; also,
remarks by H.Fried and J.Pinter. Muszaki kozl MTA 26 no.1/4:183-184
'60. (EEAI 9:10)

1. Hiradastechnikai Kutato Intezet, Ujpest (for Szep)
(Transistors) (Germanium)

H/009/61/000/004/004/005
D018/D105

AUTHOR: Fried, Henrik, Member of the Society (see Association)

TITLE: Development of the Hungarian television picture tube production and the picture tube plant at Vác

PERIODICAL: Magyar Hiradástechnika, no. 4, 1961, 158-163

TEXT: The article summarizes the main phases and aspects of experiments preceding the domestic development and mass production of picture tubes and describes the machines and equipment of the television picture tube plant at Vác, in particular the design of the conveyorized line, an original product of the "Tungsram" Plant and of the Vákuumtechnikai Gépgyár (Vacuum Machine Plant) in Budapest. The first cathode ray tubes were developed by the Távközlési Kutató Intézet (TKI) (Telecommunication Research Institute) in 1949. In 1953 this institute worked out the material for fluorescent coating consisting of zinc sulfide and zinc cadmium sulfide activated with silver. The Soviet picture tube specifications put at the disposal of Hungarian manufacturers were of little use to the designers at Vác, due to different Soviet and Hungarian types and production volume and also due to

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H/009/61/000/004/004/005
D018/D105

Development of the Hungarian television

the fact that the USSR had not yet fully developed the picture tube mass production machines at that time. The Hungarian manufacturers could only make use of the Soviet envelope washing machine. The first Hungarian picture tube developed in 1955 and based on the Soviet 31LK2B picture tube never entered mass production because it was outdated at the time its development was completed. Efforts of the "Orion" Gyár (Plant) to produce picture tubes on a large scale both for the domestic market and export failed, due to the difficulties experienced with the screen aluminizing process. The picture tube pilot plant, a section of the ME plant /Abstractor's note: ME not defined⁷ of Egyesült Izzó(United Incandescent), carried out small series production of picture tubes until the picture tube plant in Vác was put into operation in 1960. The Vác plant consists of an assembly shop, a components section, a chemical shop and a power plant. Glass envelopes are produced at the glass factory in Nagykanizsa. The metallic electronic parts, as well as the material for fluorescent coating are produced at the plant. The existing vacuum installation has 62 individual pumping units, while the second installation to be supplied by United Incandescent in Budapest will have 74 such units. The conveyorized line is

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Development of the Hungarian television

H/009/61/000/004/004/005
D018/D105

used for various picture tube manufacturing processes between the vacuum installation and the final testing section, i.e. for basing, high-voltage sparkover test, gas ratio measuring, electric cathode formation, electric parameter measuring, and illuminated screen checking. A second conveyorized line of the same design, destined for the Vác plant, is in production. There are 11 figures.

ASSOCIATION: Hiradástechnikai Tudományos Egyesület (Communication Scientific Society) and Egyesült Izzó lámpa és Villamossági Rt. (United Incandescent Lamp and Electric Company).

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FRIED, Henrik

Development of the Hungarian television picture tube manufacture and description of the factory for picture tubes in Vac. Magy hir techn 12 no.4:158-163 Ag '61.

1. Hiradastechnikai Tudomanyos Egyesulet tagja; Egyesult Izzolampa es Villamossagi Rt.

"APPROVED FOR RELEASE: 06/13/2000

CIA-RDP86-00513R000513720017-5

FRIAS, Henrik, Researcher

Role of micromanipulators in manufacturing semiconductor wafers.
Fizmekmekanika 4 no.3;87'90 Mr 165.

1. Scientific Division Chief, Research Institute of Telecommunications
Engineering Industry, Budapest.

APPROVED FOR RELEASE: 06/13/2000

CIA-RDP86-00513R000513720017-5"

FOLDES, I.:BEREGI, E.:FRIED, J.

The effect of intestinal extracts on the aorta of dogs and rabbits.
Kiserletes orvostud. 4 no. 5:356-360 Oct 1952. (CIML 23:5)

1. Doctor for Foldes and Beregi. 2. First Institute of Pathological
Anatomy and Experimental Cancer Research, Budapest Medical University.

SVOBODA, Adolf, MUDr.; FRIED, Karel, MUDr.; KUNC, Zdenek, Doc., MUDr.

Subdural arachnocele. Cesk. neur. 19 no. 3:180-184 Aug 56.

1. Nervove oddeleni OUNZ v Kladne. -Roentgenologické oddeleni
OUNZ v Kladne. Neurochirurgicke oddeleni UNZ v Praze.
(ARACHNOID, dis.
arachnocele, subdural, diag. & surg. (Cz))

FRIED, K.

SLOVANIA / Diseases of Farm Animals. Diseases Caused R-2
by Helminths.

Abs Jour: Ref Zhur-Biol., No 2, 1958, 7322

Author : K. Fried, J. Knezik, J. Duchaj, J. Jilek.

Inst : Not Given

Title : Certain Data on the Strongylidosis of Colts in Slovakia.

Orig Pub: Veterinarstvi, 1957, 7, No 4, 108-109 (Slovatsk.).

Abstract: The disease is widespread in various regions of Slovakia, and affects primarily colts three to nine weeks old. Its causal agent is Strongyloides westeri. A case is described of a mass illness of colts on one farm and the measures taken in the fight against this invasion.

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18

EJECPTA MEDICA Soc.14 Vol.12/5 Radiology May 1958
FRIED, K.

APPROVED FOR RELEASE OF 06/13/2000 CIA-RDP86-00513R000513720017-5

diagnostice mnohotetného myelomu - Fried K., Marx F. and Teichmann V. Rtg Odd. OÚNZ, Kladně; Rtg Odd. Fak. Poliklin., Praha; Rtg Odd. II Int. Klin., Praha - ČAS. LÉK. ČES. 1957, 96/31 (989-993) Illus. 8
The authors have attempted to solve several questions of importance in the X-ray diagnosis of multiple myeloma on the basis of an analysis of 42 verified observations. Their conclusions are: (1) a precise classification of the manifestations of multiple myeloma is not possible; (2) pathognomonic manifestations of myeloma in the X-ray picture do not exist, with the exception of typical destruction of the spine; (3) localization in various parts of the skeleton to a certain degree influences the character of the manifestations of myeloma. The authors' diagnostic analyses of X-ray signs is supplemented by remarks on differential diagnosis and on some of the clinical characteristics with localization of multiple myeloma in the vertebral column.
(XIV, 5, 6, 16)

FRIED, K.; PATEK, Vl.; NOVAK, V.

Considerations on the problem of Caplan's syndrome. Cesk. rentg.
15 no.1:65-68 F '61.

1. Radiologicke oddeleni OUNZ-Kladno, prednosta MUDr. Fried
Revmatologicka ordinace pri I. int. oddeleni OUNZ-Kladno, pred-
nosta MUDr. Jindrak.
(SILICOSIS radiog)
(ARTHRITIS RHEUMATOID radiog)

"APPROVED FOR RELEASE: 06/13/2000

CIA-RDP86-00513R000513720017-5

FRIED, K.

Osteochondrosis of the metatarsal joint of the toe. Caso.
radiol. 19 no.1c 41-47 Ja '65

L. Radiologicke oddeleni Obvodniho ustavu narodniho zdravi
v Kladne (vedouci ~ MUDr. K. Fried).

APPROVED FOR RELEASE: 06/13/2000

CIA-RDP86-00513R000513720017-5"

FRIED, K.

SURNAME, Given Names

Country: Czechoslovakia

Academic Degrees: Dr

Department of Clinical Diagnosis and Internal Diseases, First Veterinary

Affiliations: Faculty, Graduate School of Agriculture (Katedra klinickej diagnostiki a
vnútorných chorob I. veterinárskej fakulty VŠP /Vysoka skola polnospodarstva
Kosice

Source: Prague, Veterinarstvi, Vol 11, No 8, Aug 1961; pp 298-300

Data: "Radiologic Diagnosis of Poultry Ascariasis"

FRIED, K., Prof

JANTOŠOVIC, J.

GPO 981643

FRIED, K.;SALAYOVA, J.

Treatment of dermatitis following irradiation. Prakt. lek., Praha 33 no.
14:325-326 20 July 1953. (CIML 25:1)

1. Of the Radiological Department (Head--K. Fried, M.D.) of OUNZ Kladno.

Q-³

FRIED,K.

Management of x-ray films and contrast materials. Cesk.rentg. 14
no.5:341-346 0 '60.

1. Radiologicke oddeleni OUNZ - Kladno, prednosta MUDr, K.Fried.
(RADIOGRAPHY)

~~FRIED, K.~~

Pseudotumors of the bone. Cesk.rentg. 15 no.2:101-111 A '61.

1. Radiologicke oddeleni UJNZ Kladno, prednosta MUDr. K. Fried.
(BONE AND BONES neoplasms)
(EXOSTOSES radiography)

"APPROVED FOR RELEASE: 06/13/2000

CIA-RDP86-00513R000513720017-5

FRIED, K.

Osseous pseudotumors. Cesk. rentg. 15 no.3:167-185 '61.

(BONE DISEASES radiog)

APPROVED FOR RELEASE: 06/13/2000

CIA-RDP86-00513R000513720017-5"

FRIED, K.

A system of card-indexes of roentgen findings using the decimal system. Cesk. rentgenol. 15 no.4:284-286 '61.

1. Radiologicke oddeleni nemocnice, Kladno.
(RADIOGRAPHY) (MEDICAL RECORDS)

"APPROVED FOR RELEASE: 06/13/2000

CIA-RDP86-00513R000513720017-5

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DATE 10-10-2007 BY SP2007-311

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DATE 10-10-2007 BY SP2007-311

APPROVED FOR RELEASE: 06/13/2000

CIA-RDP86-00513R000513720017-5"

DEAK, Pal, dr.; FRIEND, Laszlo, dr.

Cistobil cholecystography. Magy. radiol. 8 no.3:163-
166 Aug 56.

1. A Peterfy Sandor utcai Korhaz-Rendelointezet (igazgato:
Lendvai, Jozsef, dr.) Rontgenosztalyanak (foorvos: Deak, Pal, dr.)
kozlemenye.

(CHOLECYSTOGRAPHY

contrast media, cistobil, comparison with sodium
iodipamide (Hun))

(CONTRAST MEDIA

cistobil in cholecystography, comparison with
sodium iodipamide (Hun))

FRIED L.

EXCERPTA MEDICA Sec.14 Vol.11/9 Radiology Sept 57.

1650. FRIED L. Röntgenabt., Krankenh. 'Péterfy Sándor', Budapest VII. *Zur Beurteilung des Zystikusstumpfes bei i.v. Cholangiographie. Demonstration of the stump of the cystic duct by intravenous cholangiography FORTSCHR. RÖNTGENSTR. 1956, 85/1 (47-54) Illus. 9

After cholecystectomy the stump of the cystic duct can be visualized by intravenous cholangiography. The size and shape of the stump are determined by (1) the surgical technique and (2) the pressure prevalent in the bile ducts. (1) If normal visualization of the common duct is associated with a larger cystic duct stump, then it can be presumed that a larger stump was left intact at operation. (2) If a dilated cystic duct stump is found in association with dilation of the bile passages and retarded evacuation, then the increased pressure in the biliary tract can be the cause of the stump dilation. (3) If dilation of the biliary tract is not associated with dilation of a cystic duct stump, then the condition involves an inflammatory dilation of the biliary tract without increased pressure. Therapy differs in the three different groups.

Bücker - Hamburg

FRIED, Laszlo, Dr.; MATYUS, Lajos, Dr.

Isolated spleen tuberculosis. Orv. hetil. 99 no.12:411-414 28 Mar 58.

1. A Fovarosi Tetenyi uti Korhaz (igazgato: Zellner Pal dr. foorvos)
Rontgenosztalyanak (foorvos: Deak Pal) es Sebeszeti Osztalyanak
(foorvos: meszaros Karoly dr.) kozlemenye.
(TUBERCULOSIS, SPLENIC, case reports
(Hun))

BALLA, Ildiko, dr.; FRIED, Laszlo, dr.

Contribution to the pathogenesis of Charcot's joint. Magy.
radiol. 11 no.4:249-253 N '59.

1. Tetenyi uti Morhas (igazgato: Zellner Pal dr.) Rontgen-
osztalyanak (foorvos: Deak Pal dr.) kozlemenye.
(JOINTS dis)

FRIED, Laszlo, dr.

The bone structure in Klippel-Trenaunay syndrome and in some other peripheral circulatory diseases. Magy. radiol. 14 no.5:257-268 S '62.

l. Az Orszagos Orvostovabbkepzo Intezet Tanszeke (Vezeto: Deak Pal dr. egyetemi tanar).

(BONE DISEASES)

MATE, Karoly, dr.; FRIED, Laszlo, dr.

On giant gastric ulcer in old age. Orv. hetil. 105 no.24:
1114-1118 14 Je'64

1. Fovososi Tanacs, Tetenyi uti Korhaz, III. Belosztaly es
Orvostorabbkepzo Intezet, Rontgenologial Tai sek.

"APPROVED FOR RELEASE: 06/13/2000

CIA-RDP86-00513R000513720017-5

IVANYI, Janos, dr.; FRIED, L.; MATE, K.

Ulcerative diseases in old age. Orv. hetil. 105 no.34:1627
23 Ag '64.

APPROVED FOR RELEASE: 06/13/2000

CIA-RDP86-00513R000513720017-5"

"APPROVED FOR RELEASE: 06/13/2000

CIA-RDP86-00513R000513720017-5

FRIED, Robert, prof. d-r

On listeriosis. Med. arh., Sarajevo 12 no. 3:57-65 My-Je '59.

1. Institut za mikrobiologiju Medicinskog fakulteta u Sarajevu.
(LISTERIA INFECTIONS)

APPROVED FOR RELEASE: 06/13/2000

CIA-RDP86-00513R000513720017-5"

FRIED, Robert, prof. d-r

Problem of intestinal bacterial flora and "dysbacteriosis".
Med. arh., Sarajevo 13 no.4:1-6 Jl-Ag '59.

1. Institut za mikrobiologiju Medicinskog fakulteta u Sarajevu,
sef: prof. d-r Robert Fried.
(INTESTINES microbiol.)

FRIED, VOJTECH

Chemical Abst.
Vol. 48 No. 8
Apr. 25, 1954
General and Physical Chemistry

Liquid-vapor equilibrium. III. Thermodynamics of
nonelectrolyte solutions. Edward Hala, Otakar Vilim,
Illi Pick, and Vojtech Fried (Vysoka škola chem., Prague,
Czech.). Chem. Listy 47, 1101-12 (1953); cf. *ibid.*
641.—A review with math. considerations and 30 refer-
ences.
M. Hudlický

8
⑤
1954/8/54

HALA, E.; FRIED, V.; PICK, J.; VILIM, O.

Equilibrium in the system liquid -- vapor. Part.4. General equation for the dependence between activity coefficients and the composition of the liquid phase. Sbor.Chekhh.khim.rab. 19 no.1:16-23 F '54. (MLRA 7:6)

1. Kafedra fizicheskoy khimii, Pravshskogo Khimicheskogo Instituta.
(Phase rule and equilibrium) (Activity coefficients)

FRIED, V.

CZECH

✓ Liquid-vapor equilibria. V. Limiting values of the relative volatilities in two-component systems at high pressures. Eduard Hala, Vojtech Fried, Jiri Pick, and Otakar Vilim. *Inst. Phys. Chem. (Prague). Collection Czechoslov. Chem. Commun.*, 19, 419-426 (1954) (in German). VI. Calculation of liquid-vapor equilibria in two-component systems from isobaric t - x curves. Eduard Hala, Jiri Pick, Vojtech Fried, and Otakar Vilim. *Ibid.* 417-27. VII. Calculation of liquid-vapor equilibria in two-component systems from isothermal p - x curves. Eduard Hala, Vojtech Fried, Jiri Pick, and Otakar Vilim. *Ibid.* 417-27. --See C.A. 48, 4301a&c. R. I. C.

Fried, Vojtech

CZECH

Liquid-vapor equilibria. VIII. A new flow equilibrium still for the determination of liquid-vapor equilibria. Otakar Vilim, Eduard Hala, Vojtech Fried, and Jiri Pick. Collection Czechoslov. Chem. Commun., 19, 1330-4 (1954) (in German).—See C.A. 48, 4301d. E. J. C.

Friedl Kvitex

Liquid-vapor equilibria. X. The system butanol-butyl
ether-butyl methacrylate at low pressures. Vojtěch Erdoš,
Jiří Pick, Eduard Hala, and Otakar Vilím (Výzkumný ústav
chemie, Praha, Czech.). Chem. Listy 48, 181-7 (1954);
cf. C.A. 48, 4301e.—The isothermal equil. compns. of the
vapor and liquid phases at 65° in the 3 binary systems were
computed from the dependences of vapor pressures on
temp. and compn. The relative volatilities in the ternary
system were calc'd. by the modified two-suffix Scatchard
equation [C.A. 30, 3581] from the consts. of binary systems.
E. Erdős ✓

FRIED, V.

Vapor pressures of butyl α -hydroxybutyrate and of
diethyl ether. V. FRIED, J. PEC, E. HATA, and O. VILIM

(Vysoké Školy chem. Praha, Czech.J. Chem. Listy 48,
774-8(1954).—From osmometric measurements the temp.
dependence of vapor pressure of Bu α -hydroxybutyrate
(I) and of Bu₂O was computed in the form: $\log \mu = A -$
 $B/(T - 43)$, where for I: $A = 7.6476$, $B = 1931.0$ and
for Bu₂O: $A = 7.4839$, $B = 1711.5$; both in the range
70-760 mm. Hg. The phys. consts. found were: I b.p.
185.0°, d₄₀²⁰ 0.95035, n_D²⁰ 1.41, I₁ Bu₂O b.p. 143.2°, d₄₀²⁰
0.70843, n_D²⁰ 1.3900.

FRIED, V.

5(4)

PHASE I BOOK EXPLOITATION

CZECH/2501

Hala, Eduard, Jiri Pick, Vojtech Fried, and Otakar Vilim

Rovnovaha kapalina--para (Liquid--Vapor Equilibrium) Praha, Nakladatelstvi Ceskoslovenske Akademie Ved, 1955. 321 p. (Series: Ceskoslovenska Akademie ved. Studie a prameny. Sekce chemicka, sv. 10) Errata slip inserted. 38,600 copies printed.

Scientific Ed.: Jan Pinkava, Doctor, Engineer; Resp. Ed.: Jaroslav Vacha, Doctor.

Full English translation under the title Vapor-Liquid Equilibrium.
[Translator: G. Standart] published in 1958 by Pergamon Press Ltd.
Library of Congress call number: TP156.E65R613.

Card 1/1

41S/gmp
11-10-59

ENR'D, V

CZECH

Packings for laboratory fractionating columns made from glass textiles. I. Hala, O. Vilim, J. Pick, and V. Fried (Vysoka škola chem. technolog., Prague). *Chem. Listy* 49, 350-353 (1955).—Three types of packings are described: (a) by the use of a helical screw from glass textile tube, a HETP of 1.8-3.6 cm. has been obtained. (b) In using "heligrid" type of packing from steel wire spiral, better contact with the walls and higher efficiency (HETP = 1.4 cm.) has been achieved by inserting suitably made disks from glass textile. (c) The efficiency of current packing is increased three-fold, if sepd. by appropriately perforated glass textile disks.

62

FRIED, V.

7

③ ✓ Liquid-vapor equilibria. XIII. Phase equilibria in the system water-butanol-butyl acetate. Jiri Pick, Vaclav Fried, Eduard Hala, and Otakar Vilim (Vysoka škola chem.-technol., Prague). *Chem. Listy* **49**, 1112-16 (1955); cf. *C.A.* **49**, 7061. — The liquid-liquid equil. in the ternary system H₂O-BuOH-AcOBu was detd. by the synthetic method; the temp. dependence of the equil. curve was obtained by graphic correlation. The tie-lines were detd. analytically. The vapor-liquid equil. in the homogeneous region of the ternary system was calcd. by the 3-suffix van Laar equation; the ternary consts. were evaluated from 2 exptl. points detd. analytically. E. Erdos

"APPROVED FOR RELEASE: 06/13/2000

CIA-RDP86-00513R000513720017-5

6

1. Initial system configurations. XIII. Phase 1
of the system water treated butyl acetate. On Page 1
of the document, Initial Name and Unit,
the following information is contained:

APPROVED FOR RELEASE: 06/13/2000

CIA-RDP86-00513R000513720017-5"

Lived, V.

6

Vapor pressure of ethylene glycol monoinethyl and monoethyl ethers. J. Pick, V. Fried, E. Hala, and O. Vilim.
Collection Czechoslovak. Acad. Sci., Prague, 21, 260-1 (1950) (in
German). See C.A. 50, 636h.

4

PM rank

CZECHOSLOVAKIA/Thermodynamics. Thermochemistry. Equilibria. Physics- E-8
Chemical Analysis. Phase Transitions.

Abs Jour : Ref Zhur - Khimiya, No 8, 1957, 26107

Author : Eduard Hala, Vojtech Fried, Jiri Pick, Otokar Vilim
Title : Equilibrium Liquid - Vapor. XIV. Activity Factors and Physical
Properties of Pure Components.

Orig Pub : Chem listy, 1956, 50, No 3, 343-348

Abstract : The authors proposed a new method of computation of the dependence of the activity factor on the composition of the liquid mixture. This method always permits to carry out the computation for a complete group of substances on the basis of the known behavior of standard binary systems and parachors of given components. Following relations were deducted basing on certain assumptions: $A_{ik}^{0.5} = A_{ij}^{0.5} - K N_j (N_j - N_k) / (N_i N_k)$ and $A_{ik}^{0.5} = A_{ij}^{0.5} - K' (N_j - N_k) / \bar{P}_i$, where A_{ij} , A_{ik} , \bar{P}_i and A_{ki} are constants of Van Laar equations of the 3rd order for binary systems ij and ik , and K and K' are constants which it is necessary to determine for the given group of binary mixtures. The magnitudes of N_i , N_j and N_k are given by the relation $N_i = (0.377 \bar{P}_i + 11.0)0.625$, where \bar{P}_i is the parachor of the i -th component. See RZhKhim, 1956, 77532 for the report XIII.

Card : 1/1

Fred, Vogtlich

9

and paper conditions via analysis of
the physical properties of paper samples.

(P60) If $T_1 = 40$ minutes, $\Delta T_1 = 10$ minutes, and $\Delta T_2 = 10$ minutes, we can write the following linear equations for A_1 and A_2 in terms of N_1 , N_2 , N_3 , N_4 , K_1 , and K_2 :

$$N_1 A_1 + N_2 A_2 = K_1^2 K_2^2 N_1 N_2 N_3 N_4$$
$$N_1 A_1 + N_3 A_2 = K_1^2 K_2^2 N_1 N_2 N_3 N_4$$

where A_1 and A_2 values are measured in percent loss of weight per hour, N_1 , N_2 , N_3 , and N_4 values are measured in percent loss of weight per minute, K_1 and K_2 are constants characteristic of the given group of paper systems.

FRIED, VOLKMAR

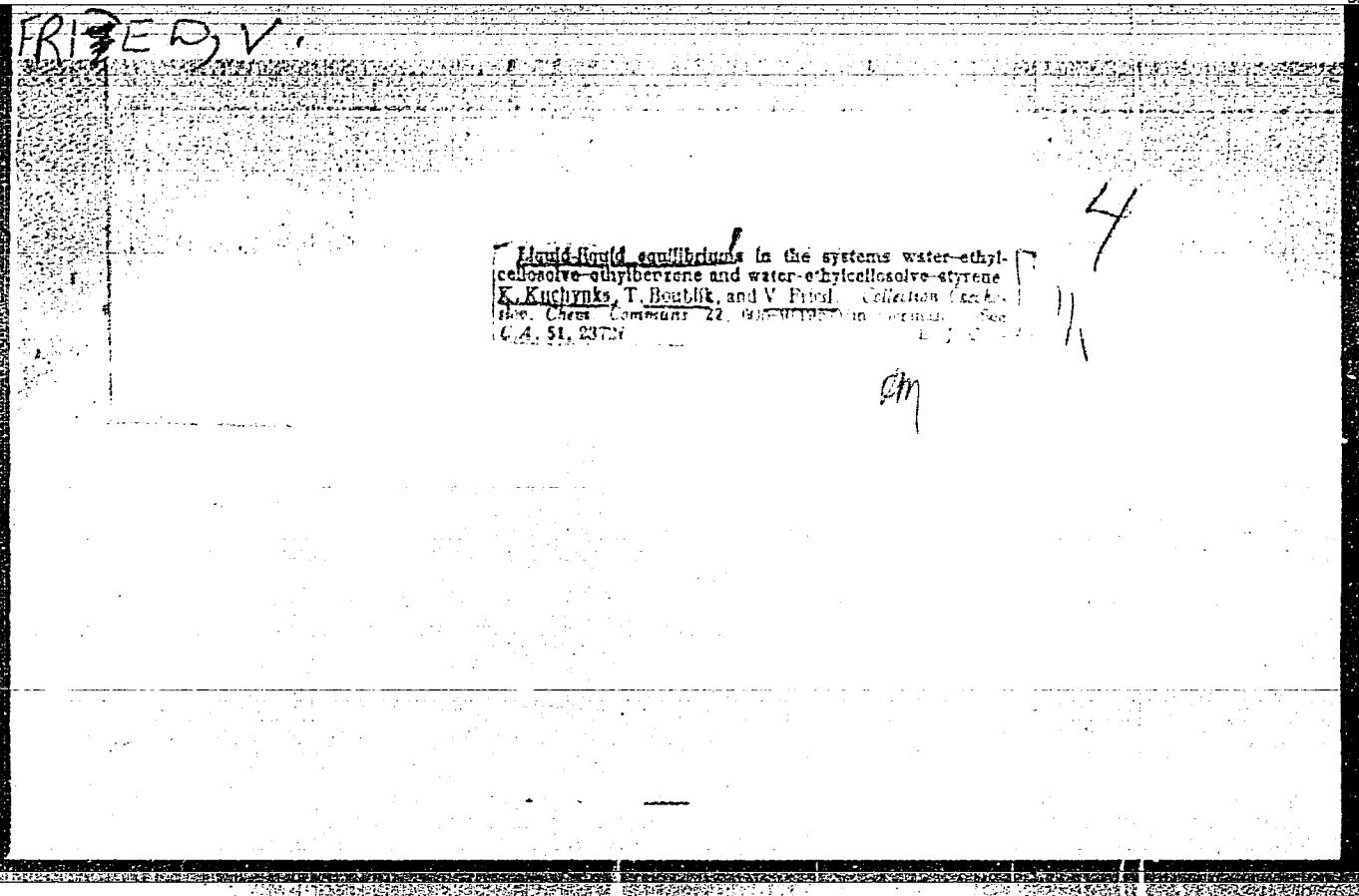
Phase equilibria in the systems 3-methoxy-1-methylbenzene + styrene and 2-methoxyethanol + water. Part I.

Johannick, Volkmar Fried, and Josef Kipp. Institut für Physikalische Chemie der Universität Regensburg, D-8400 Regensburg, FRG.

The ternary system benzene + styrene + $\text{CH}_3\text{OCH}_2\text{CH}_3$ were studied at 25°C by Raoult's law and van Laar's equation of state. $A_{12} = 0.003$, $A_{13} = 0.001$, $A_{23} = 0.001$. The result in the form of a phase diagram at 25°C is shown below. The data at 0.01 mm Hg are: 1 mol. % of I, 51.0% of II, 48.0% of III; 1 mol. % of I, 51.9% at 0.01 mm Hg; 1 mol. % of I, 51.0% at 0.02 mm Hg; 1 mol. % of I, 51.5% at 0.05 mm Hg; 1 mol. % of I, 51.7% at 0.1 mm Hg; 1 mol. % of I, 51.8% at 0.2 mm Hg; 1 mol. % of I, 51.9% at 0.5 mm Hg; 1 mol. % of I, 52.0% at 1.0 mm Hg. No azeotrope was found. The results are discussed in connection with the isolation of III.

"APPROVED FOR RELEASE: 06/13/2000

CIA-RDP86-00513R000513720017-5



APPROVED FOR RELEASE: 06/13/2000

CIA-RDP86-00513R000513720017-5"

CZECHOSLOVAKIA / Physical Chemistry. Thermodynamics. B
Thermochemistry. Equilibria. Physico-
Chemical Analysis. Phase Transitions.

Abs Jour: Ref Zhur-Khimiya, No 17, 1958, 56746.

Author : Jakubicek Josef, Fried Vojtech, Vahala Josef.

Inst : Not given.

Title : Phase Equilibria in the Systems 2-Methoxyethan-
ole - Ethyyl-Benzene - Styrene and 2-Methoxy-
ethanole - Aqua.

Orig Pub: Chem. listy, 1957, 51, No 8, 1422 - 1428.

Abstract: Attempting to separate styrene and ethylbenzene,
occurring in a liquid dehydrogenation product,
the authors investigated the liquid - vapor
equilibrium at a pressure of 62 millimeters of
the mercury column by applying 2-methoxyethan-
ole (methylcellosolve, I) as a third component.

Card 1/5

12

CZECHOSLOVAKIA / Physical Chemistry. Thermodynamics. B
APPROVED FOR RELEASE: 06/13/2000 CIA-RDP86-00513R000513720017-5
Chemical Analysis. Phase Transitions.

Abs Jour: Ref Zhur-Khimiya, No 17, 1958, 56746.

Abstract: The data on the systems ethylbenzene - styrene,
I - ethylbenzene and I - styrene were submitted.
The constants in the Van Laar equations were
calculated for all the systems. The behavior
of the system ethylbenzene - styrene is prac-
tically ideal. The constant C for the tertiary
system has been calculated from the binary sys-
tem constants by applying the Vol Law: $C = 0.5$
 $(0.591 - 0.755 \neq 0.475 - 0.685) - 0.187$. The
relative volatility values have been calculated
from the activity coefficient, and from the
former — the equilibrium composition of the liq-
uid and vapor phases of the tertiary system. The
calculations have been verified by measurements.

Card 2/5

CZECHOSLOVAKIA / Physical Chemistry. Thermodynamics. B
Thermochemistry. Equilibria. Physico-
Chemical Analysis. Phase Transitions.

Abs Jour: Ref Zhur-Khimiya, No 17, 1958, 56746.

Abstract: I - ethylbenzene possesses an azeotropic point; The composition of the azeotropic mixture at 62 millimeters of the mercury column is 42.1 mole percent I, and the boiling point 51.9°C. The system I - styrene forms an azeotrope with a composition of 57.9 mole percent I and a boiling point 56.8°C, at the same pressure. No tertiary azeotrope was observed. The azeotropic mixture I - ethylbenzene was obtained on a test column (25 theoretical plates) by continuous vacuum purification; the I - styrene mixture was taken from the hot-water boiler. The system H₂O - I forms an azeotropic mixture with

Card 4/5

Physical Chemistry. Thermodynamics.
Thermochemistry. Equilibria. Physico-
Chemical Analysis. Phase Transitions.

Abs Jour: Ref Zhur-Khimiya, No 17, 1958, 56746.

Abstract: 94.5 mole percent H₂O, boiling point 99.2°C at 752 millimeters of the mercury column, 98.7 mole percent H₂O and boiling point 51.5°C at 100 millimeters of the mercury column. The application of I as a third component permits the reduction of the number of plates, necessary for the separation of styrene, which is reduced from 36 to 20, but which however, complicates the separation process.

Card 5/5

CZECHOSLOVAKIA / Physical Chemistry. Thermodynamics. B
Thermochemistry. Equilibris. Phase
Changes. Physico-chemical Analysis.

Abs Jour : Ref Zhur - Khimiya, No 12, 1959, No. 41572

of the above expression may be expressed by the following type of equation:
 $\ln \eta_E = X_1 X_2 [b + c(X_1 - X_2) + d(X_1 - X_2)^2 + ..]$
where b, c and d are constants which can be determined from the experimental data.
The number of necessary constants may be found from a plot of $\ln \eta_E$ vs $X_1 X_2$. When the number of constants is equal to zero, the system behaves ideally. From cited examples, the system CCl_4 -benzene is an ideal one, while the benzene-cyclohexane system may be described by means of one constant. Two constants were utilized for

Card 2/3

CZECHOSLOVAKIA / Physical Chemistry. Thermodynamics. B
Thermochemistry. Equilibris. Phase
Changes. Physico-chemical Analysis.

Abs Jour : Ref Zhur - Khimiya, No 12, 1959, No. 41572

a mixture of acetone and ethanol. Three constants were required for the system CH_3OH -dichloroethane. A relationship between the excess viscosity and excess free enthalpy, ΔG_E : $-\ln \eta_E = \Delta G_E / 2.45 RT$, was deduced from Eyring's Theory. It is presumed that analogous methods may be applied to 3-component systems.
-- O. Knessl

Card 3/3

Country : Czechoslovakia B-8
 Category : Thermodynamics. Thermochemistry. Equilibria.
 Abs. Jour. : Physico-Chemical Analysis. Phase Transitions.
 Ref Zhur-Khimiya, No 6, 1959 18436

Author :
 Institut. :
 Title :

Orig. Pub. :

Abstract : system; ternary constant was determined from data on liquid-vapor equilibrium of ternary system in the region where the influence of the term of Van Laar equation containing this constant is greatest (region wherein concentrations of two components are about equal and concentration of third component is low). A rapid method is described for calculating the activity coefficient in ternary system. Communication XVIII see RZhKhim, 1957, 53896. -- O. Knessl.

Card: 2/2

B-11

APPROVED FOR RELEASE 06/13/2000 CIA-RDP86-00513R000513720017-5-11
 COUNTRY : Czechoslovakia
 CATEGORY : Theories of Acids and
 Bases.
 ABS. JOUR. : RZhKhim., No. 23 1959, No. 21491
 AUTHOR : Fried, V.; Hala, E.; Pick, J.
 INST. : Not given.
 TITLE : Viscosity of the Nonelectrolytic Solutions.
 ORIG. PUB. : Collect. Czechosl. Chem. Comms, 1959,
 24, #2, 400-404.
 ABSTRACT : See RZhKhim, 1959, #12, 41572.

CARD: 1/1

FRIED, V.

Distr: 4E2c(j)/4E3d

✓ Solubility of acetylene in benzene, toluene, and p-xylene. 1-OK (CW)
J. Vitovec and V. Fried (Vyzkumný ústav syntetického
kaucuku) Gottwaldov, Czech.) Collection Czechoslov.

Chem. Commun. 25, 1652-6(1960).—The solv. of acetylene
in benzene (I), toluene (II), and p-xylene (III) was detd. by
the statn. method in the temp. range from 20° to: for I 60°,
II 90°, and III 120°. The results were expressed by the
Bunsen absorption coeff. α : $\log \alpha = (A/T) + C$, where the
consts. A and C have the values: for I 626.00, -1.4600;
II 665.81, -1.6480, III 630.06, -1.5628. The results
agree well with the Hildebrand solv. theory; the mean
solv. parameter of acetylene is 6.80. The ratio of abs.
temps. at which the gas has the same solv. in 2 solvents is
const. E. Erdélyi

4

2

Distr: 4E2c(j)/4E3d

✓ Liquid-vapor equilibria. XXIII. The phase equilibria in the binary systems vinylacetylene-benzene, vinyl-acetylene-toluene, and vinylacetylene- α -xylene. V. Fried and J. Vitovcev (Vysoká škola chemicko-technol., Prague). Collection Czechoslov. Chem. Commun. 25, 1622-8 (1960); cf. CA 54, 10068s. — The isobaric vapor-liquid equil. were detd. at 740 mm. Hg on a modified circulation app. and correlated by means of the Hildebrand 3-const. equation (CA 53, 21107b). The agreement with the Hildebrand solv. theory is good; the behavior of the systems studied does not deviate significantly from an ideal one. E. Erdős

AMT

6
1-BW (BW)
2-NAT (NB) (MAY)

2

FRIEDL V.

Distr: 4E2c(j)/4E3d

✓ Solubility of vinylacetylene in benzene, toluene, and
p-xylene. L. Vlčová and V. Fried (Výzkumný ústav
synthetického řezu, Gottwaldov, Czech.). Collection
Československých Časopisů, 25, 2218-21 (1960) (in German).

The solv. of the vinylacetylene in benzene, toluene, and
p-xylene was detd. by the satn. method. These systems
do not obey Henry's law. The results are expressed by the
equations: C₆H₆, log x = (1277.9/T) - 2.6170; toluene, log
x = (1233.3/T) - 2.6356; p-xylene, log x = (1165.8/T)
- 2.2443, where x denotes the solv. expressed by the
mole fraction at the partial pressure 760 mm. Hg and T
is the abs. temp. The results of the satn. and of the
circulation method (C4 84, 20450a) agree. E. Brdo

4
BW(BW)
JAJ(NB)
2

VITOVEC, J.; FRIED, V.

Solubility of acetylene in benzol, toluol, and p-xylene. Coll Cs
Chem 25 no.6:1552-1556 Je '60. (EEAI 10:9)

1. Forschungsinstitut fur synthetischen Kautschuk, Gottwaldov und
Institut fur physikalische Chemie, Technische Hochschule fur Chemie,
Prag.

(Acetylene) (Benzene) (Toluene) (Xylene)

FRIED, V.; VITOVEC, J.

Lique-steam equilibrium. XXXII. Phase equilibrium in vinylacetylene-benzene, vinylacetylene-toluene, and vinylacetylene-p-xylene.
Coll Cz Chem 25 no.6:1642-1648 Je '60. (EEAI 10:9)

1. Institut fur Physikalische Chemie, Technische Hochschule fur Chemie,
Prag, und Forschungsinstitut fur synthetischen Kautschuk, Gottwaldov.

(Butyne) (Benzene) (Toluene) (Xylene)
(Phase rule and equilibrium)

FRIED, V. i PICK, J.

Liquid vapor equilibria. Part 25: System phenol-acetophenone at reduced pressure. Coll Cz Chem 26 no.4:954-960 Ap '61.

1. Department of Physical Chemistry, Institute of Chemical Technology, Prague.

(Vapors) (Phenol) (Acetophenone)

NOVAK, J.; FRIED, V.; PICK, J.

Solubleness of carbon dioxide in water under various pressures and temperatures. Coll Cz Chem 26 no.9:2266-2270 '61.

1. Institut fur physikalische Chemie, Technische Hochschule fur Chemie, Prag.

(Carbon dioxide)

2
CZECHOSLOVAKIA

VESELY, F; FRIED, V; PICK, J.

Institute of Physical Chemistry of the Technical High School
of Chemistry, Prague (foi all)

Prague, Collection of Czechoslovak Chemical Communications,
No 6, 1963, pp 1459-1466

"Equal Weight Fluidity-Fluidity in the System n-Butylacetate-
Water-Phenol."

FRIED, V.; CAPKOVA, A.; SUSKA, J.

Liquid-vapor equilibrium. Pt. 31. Coll Cz Chem 28 no. 12:
3171-3179 D '63.

1. Institut fur physikalische Chemie, Technische Hochschule
fur Chemie, und Institut fur theoretische Grundlagen der
chemischen Technik, Tschechoslowakische Akademie der
Wissenschaften, Prag.

CAPKOVA, A.; FRIED, V.

Liquid-vapor equilibrium. Pt.30. Coll Cz Chem 28 no.8:2235-2239
Ag '63.

1. Institut fur theoretische Grundlagen der chemischen Technik,
Tschechoslowakische Akademie der Wissenschaften und Institut fur
physikalische Chemie, Technische Hochschule fur Chemie, Prag.

CAPKOVA, A.; FRIED, V.

Liquid-vapor equilibrium in the tetrachlorosilane-trimethylchlorosilane system. Coll Cz Chem 29 no.2:336-340 F '64.

1. Institute of Theoretical Principles of Chemical Technology,
Higher School of Chemical Technology, Prague.

LINEK, J.; FRIED, V.; FICK, J.

Liquid-vapor equilibrium. Pt.34, Coll. Czech. Chem. Soc. no.5;1358-1365
My '65.

I. Institute für physikalische Chemie, Hochschule für Chemie,
Prague. Submitted October 22, 1964.

FRIED, Vaclav

Distri: 4E2c(j)

7
2M
1
Liquid-vapor equilibria. XIX. Phase equilibria in
the system propanol-water-propyl acetate. Jiri Frik,
Edward Hala, and Vaclav Fried (Vysoká škola chem.
techn., Prague). Chem. Listy 52, 561-6 (1958); C.A. 51:

23495. — The binodal curve at the normal b.p. and the de-
pendence of the b.p. on the compn. of the homogeneous
liquid were detd. ebulliometrically in the system PrOH-
H₂O-PrOAc. In the binary system PrOH-PrOAc the
vapor-liquid equil. was detd. and the relative volatility was
correlated empirically. The vapor-liquid equil. in the H₂O-
PrOAc system was calc'd. from the mutual solv. at the b.p.
From these and published data on the system PrOH-H₂O,
the consts. of the van Laar 3-suffix equations for the ternary
system were calc'd.; the ternary const. was calc'd. from the
vapor-liquid equil. measurement on the ternary system.
A y-x diagram of the ternary system is given. The most
suitable concn. range for the detn. of the ternary const. is
discussed.

J.F.
E.Fried

"APPROVED FOR RELEASE: 06/13/2000

CIA-RDP86-00513R000513720017-5

FRIEDBERGER, R. - Strojirenstvi - Vol. 5, no. 2, Feb. 1955.

Development of cyclone furnaces. p. 100.

SO: Monthly list of East European Acquisitions, (EHAL), LC, Vol. 4, No. 9, Sept. 1955
Uncl.

APPROVED FOR RELEASE: 06/13/2000

CIA-RDP86-00513R000513720017-5"

FRIKDBERGER, V., dr.; MARYSKA, J., dr.; ZAHOUREK, V., doc. dr.

Rupture of the bladder in fractures of the pelvis. Cas.lek.cesk.
91 no.41:1183-1186 10 Oct 52.

1. Z II. chirurgicke kliniky Karlovy university (prednosta prof.
dr. Jiri Davis) a z jejeho roentgenologickeho odd. (ved. lekar
doc. Zahourek).

(FRACTURES,
pelvis, with bladder rupt.)
(PELVIS, fractures,
with bladder rupt.)
(BLADDER, rupture,
in pelvis fract.)

FRIEDBERGER, V., MUDr.; NECHANICKY, R., MUDr.

Fractures of the sternum. Acta chir. orthop. traum. czech. 23
no.3:121-123 June 56.

1. Z II. chirurgicke kliniky akademika J. Divise.
(STERNUM, fract.
management (Cz))
(FRACTURES,
sternum, management (Cz))